

**Cruise report 09A12, R/V Dana, Hirtshals-Reykjavik,  
27/8-2/9 2012**

**Oceanography at Sea DTU course 25501.**

**Colin A. Stedmon, DTU Aqua  
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### **Scientific Participants**

Colin Stedmon (Cruise leader, Course coordinator)

Andre Visser (Teacher)

Stiig Markager (Teacher)

Katherine Richardson (Teacher)

Tommy Nielsen (Technician)

Nikolaj Thyssen Dam

Lilli Gruwier Larsen.

Karl-Søren Geertsen

Thomas Bech-Thomassen

Ciaran Joseph Murray

Christina Søegren

Stavroula A. Tsoukali

Anette Christensen

Søren Enghoff-Poulsen

Arief Rullyanto

Sanne Andrén

Jakob Thyrring Kristiansen

Mette Vodder Carstensen

Haidi Cecilie Petersen

Ida Margrethe Ringgaard

Susan Guldborg Graungård Petersen

Nikolaj Sørensen

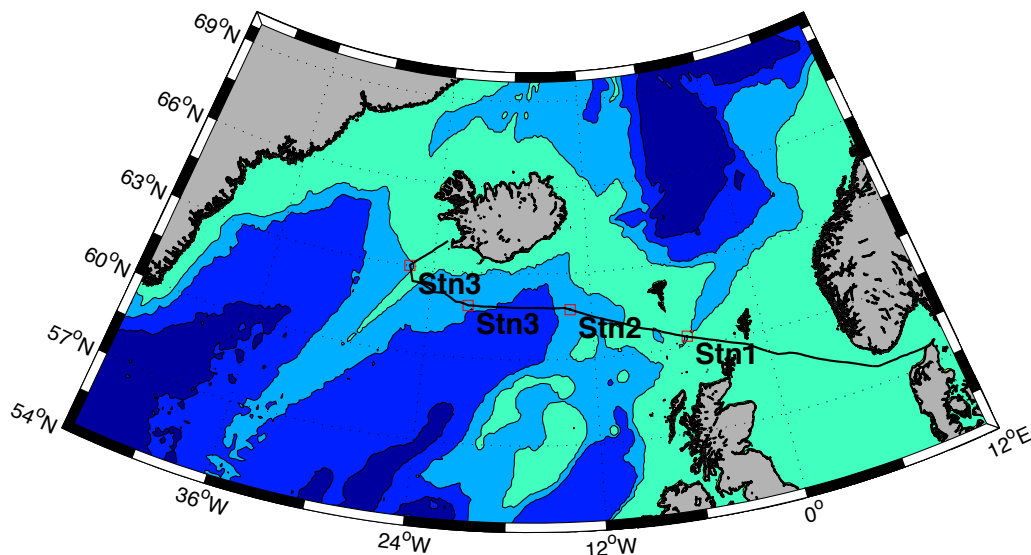
Nanna Finne Jensen

### Cruise Objective

This cruise was the focus of a university course (5 ECTS) in oceanographic research techniques on board R/V Dana, sponsored by the Danish Centre for Marine Research. The course was designed to give students sea-going experience and a practical introduction to ocean sampling. The learning objectives of the course were:

- Identify common pitfalls and necessities with organizing oceanographic sampling
- Use a conductivity-temperature-depth (CTD) probe for measuring the physical properties of seawater
- Conduct measurements of underwater light penetration and assess the contribution of different water constituents to light attenuation.
- Quantify phytoplankton biomass distribution using chlorophyll a measurements and assess phytoplankton productivity.
- Conduct measurements of chemical properties of seawater (concentrations of oxygen and selected nutrients)
- Use ODV for processing and presenting oceanographic data.
- Integrate and interpret the oceanographic data collected.
- Relate the data collected during the cruise to regional oceanographic phenomena.

**Figure 1: The route sailed and the location of the stations visited. At each station CTD (+light) profiles and water samples were collected.**



### Nature of work carried out on board.

The course was designed to have four oceanographic sampling themes: biological; physical; chemical and optical. The students were split into four groups and rotated between each theme each day. Biological oceanographic theme covered sampling

for phytoplankton and measuring chlorophyll concentrations using a spectrophotometer and carrying our Winkler oxygen measurements. Physical measurements included operating the CTD, coordinating water sampling between the groups, communicating with the bridge, and sampling for salinity calibrations. The chemical oceanography components covered sampling and analysis of pH, Alkalinity, phosphate and silicate concentrations, and the optical theme covered measurements of the attenuation of broadband light (PAR) in the surface waters and comparing these with the optical properties of particles and dissolved organic matter measured from water samples.

### **Cruise log**

27<sup>th</sup> August,

All arrived in Hirtshals and boarded Dana. Equipment set up and secured in the laboratories. Sailed from Hirtshals around 14:00 (UTC) in good weather.

Course started with a brief introduction and outline by Colin. After dinner, Andy gave a talk introducing some of the fundamentals of ocean physics. At around midnight the sea state began roughen and winds picked up.

28<sup>th</sup> August,

The main plan for the day was lectures while we steamed across the North Sea towards the first sampling station south of the Faroe Islands. There were high winds and 5 m swells making for challenging teaching conditions. During the day the conditions improved. During the morning introductory lectures on aspects of biological, chemical and optical oceanography were given by Katherine, Colin and Stiig, respectively. After lunch an introduction to the oceanography software Ocean Data View was given and the students carried out some exercises with the software.

29<sup>th</sup> August,

Swells picked up again a little in the night but by morning returned to relatively calm and overcast conditions. The morning was used to prepare for the first sampling in groups. We arrived at Station 1 in the Faroe-Scotland Channel at about 13:30 and started with a shallow CTD cast, collecting water for phytoplankton abundance and primary production, and bio optics measurements. Then carried out a deep CTD cast down to 1100 m, collecting water for water chemistry. Students worked on the samples and data collected while the ship then continued to steam west towards the Iceland Basin. It was a late night in the lab for some and a long night processing CTD data for others.

30<sup>th</sup> August,

Sea state fine, only a gentle swell, and the clouds where clearing. Arrived at Station 2 in the Iceland Basin at around 9 am. Carried out two casts again, a deep and a shallow. Students worked on their samples and data during the rest of the day. After dinner two Ph.D. students, Areif (DTU) and Ciaran (AAU) gave talks presenting their PhD projects.

31<sup>st</sup> August,

Arrived early in the morning at the third station south of Iceland. First cast with the CTD started at 6 am down to approximately 1000 meters. Followed by a shallow cast for phytoplankton samples and light measurements. Students busy the rest of the day in the lab although we were finished earlier due to the early start. Before dinner Nikolaj Sørensen (KU) gave a presentation of his PhD and after dinner Stavroula (DTU) presented her project. A whale was spotted in the distance at about 8 o'clock in the evening and everyone was up in the bridge on the look out. It did not come too close and it was difficult to see its body or tail as there was quite a swell. In the evening the weather turned bad again with high winds and swell. A gentle reminder for the students that anything not strapped down or correctly stowed away goes flying!

1<sup>st</sup> September

Arrived on station southwest of Iceland at 5:30 in the morning. Same sampling plan carried out as the day before. In the afternoon, whilst steaming to Reykjavik, the students were then reorganized into the groups they will be working up data in, and began discussing with the teachers how to present and interpret the data and plans for eventually making a poster. After dinner, Stiig gave a short talk that he was preparing for the Danish parliament on the status of nitrogen loadings to Danish coastal waters. After the talk we arrived in Reykjavik. The students continued in their groups working on the data collected from the cruise. Andy Visser disembarked.

2<sup>nd</sup> September

After breakfast, a short meeting and group picture, the students disembarked, tired but happy. The teaching cruise appeared to have been a success, with the students grasping the realities of sampling at sea and obtaining hands on experience with a range of techniques.

### **Findings.**

To document the findings of the work the four posters created by students are in the appendix of this report.

### **Concluding remarks.**

The course was a great success. The crew of Dana, the teachers and the students all enjoying the experience. I would like to take the opportunity to thank all involved and especially the Danish Centre for Marine Research for providing the funds to support the course. The teaching on board worked well and I highly recommend that this is repeated in the future. Dana and its crew provide an inspirational teaching and learning environment.

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# Biological oceanography at Sea

## Aim

The aim of this cruise was to learn to organize biological oceanography data sampling. A conductivity-temperature-depth (CTD) probe measured physical, chemical, optical and biological properties of North Atlantic water and we used the sampling results to quantify phytoplankton biomass distribution and productivity. Some CTD-plots are presented in the software program OceanDataView (ODV). We focus on the different factors influencing phytoplankton productivity and distribution and in addition discuss the changing conditions in the world's oceans.

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H. Cecilie Borgsted Petersen<sup>2</sup>  
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Dansk Center for Havforskning



DTU Aqua  
National Institute of Aquatic Resources

AARHUS UNIVERSITY

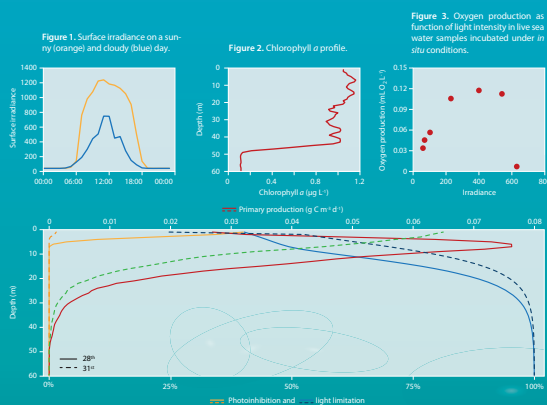


Figure 4. Primary production (red) in the water column on a cloudy and sunny day, and the relative importance of photoinhibition (orange) and light limitation (blue), over a 24-hour period.

## Primary production

The light regime may change considerably on a small scale time. The surface irradiance measured on a cloudy and sunny day here are only three days apart in the North Atlantic (fig. 1).

Recent windy conditions had likely mixed the upper layer, removing any spatial variation with depth but also between stations. There was still stratification however, ensuring a lower layer with few to no algae (fig. 2).

As irradiance increases, so does primary production. If the light intensity becomes too high however, the algae will take damage, severely reducing their photosynthesis; this is called photoinhibition (fig. 3).

Based on light profiles from a sunny and cloudy day, incubation experiments at different light intensities and a chlorophyll a profile (to determine the autotrophic biomass), the primary production at each depth could be calculated over a 24-hour period (fig. 4).

Photoinhibition only plays a role in the uppermost layer on a sunny day. Light limitation is most pronounced at depths, but also plays a role in the upper layers due to low light at night. Highest primary production is found on sunny days ( $1.01 \text{ g C m}^{-2}$ ) compared to cloudy days ( $0.69 \text{ g C m}^{-2}$ ).

On sunny days, highest primary production is seen at some depth due to photoinhibition. On cloudy days, irradiance is not sufficient to cause photoinhibition why light is almost always limiting. Here, highest primary production is seen at the surface (fig. 5, 6 and 7).

## Changing ocean conditions

When ocean conditions are changed, by natural or anthropogenic causes it has an effect on the biological pump and the organisms that are part of it.

Increasing atmospheric concentrations of carbon dioxide is reducing the pH of the oceans (by increasing hydrogen ion concentrations), making the sea more acidic. The concentration of calcium carbonate is also decreasing, and this could potentially be detrimental to calcifying organisms, such as corals and some forms of plankton, making them particularly vulnerable to climate change. Because these organisms are part of the biological pump and because  $\text{CaCO}_3$  acts as ballast to sinking material, this could potentially have a large impact on primary production and deep-sea export of DOM.

Warmer temperatures will limit the supply of nutrients in the surface layers, and that will decrease primary production in the euphotic zone. Often, production in surface layers are limited by nutrient supply from below and an increase in surface layer stratification will have a negative effect on nutrient supply.

Our growing understanding of how our world and seas is working, needs to be passed on to the people who live here, without whom we have no chance of stopping this potentially detrimental effect.

## Acknowledgement

This work is part of the marine field course Oceanography at Sea launched by Dansk Center for Havforskning. Associate Professor Colin A. Stedmon (DTU-Aqua), Professor André Viser (DTU-Aqua), Professor Katherine Richardson (Copenhagen University) and Professor Stig Markager (Aarhus University). Thanks to the Crew at Dana for taking such good care of us!

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Sabine, C.L. & Tanhua, T. (2010). Estimation of Anthropogenic  $\text{CO}_2$  Inventories in the Ocean. - Annual Review of Marine Science 2:175-198.  
Sverdrup, H.U. & Armstrong, E.V. (2009). The Chemistry of Seawater. In: An introduction to the World's Oceans, 10<sup>th</sup> ed., pp. 148-166. - McGraw-Hill Custom Publishing.  
Vanni, M. (1996). The Marine Carbonate System. In: C.P. Semmler & S.A. Thorne. Oceanography: An Illustrated Guide, pp. 182-194. - London: Manson Publishing Ltd.

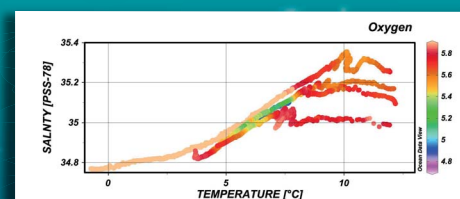


Figure 5. Salinity, temperature and oxygen plot.

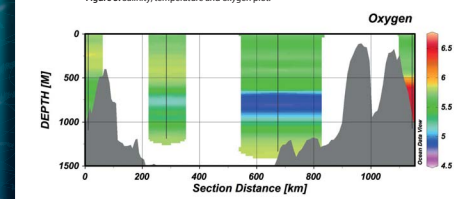


Figure 6. Oxygen profile as a function of distance.

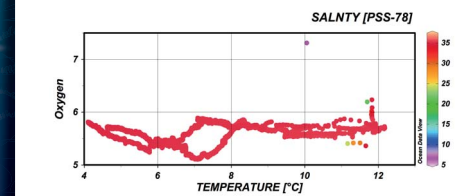


Figure 7. Oxygen, temperature and salinity plot.

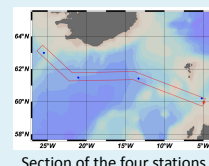
# Estimation of Anthropogenic CO<sub>2</sub> in the North Atlantic

Dansk Center for Havforskning, Dana, summer 2012.  
Christina Sægren, Nikolaj Thyssen Dam, Sanne Høj André, Stavroula Tsoukali, Thomas Bech-Thomassen.



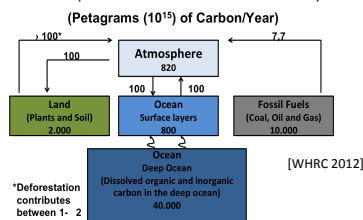
## Introduction

During the course 'Oceanography at sea' on board Dana, we examined different chemical properties at four different stations in the North Atlantic. We measured the pH, alkalinity, phosphate and silicate. We used these measurements to estimate the anthropogenic CO<sub>2</sub> in the water column.



### Global flows of carbon

The ocean plays a dominant role in the Earth's carbon cycle. The figure shows the global carbon cycle, which involves the carbon in and exchanging between the Earth's atmosphere, fossil fuels, the oceans (the surface layers and the deep ocean), the vegetation and soils of the terrestrial ecosystems. The oceans constitute the largest active pool of carbon at the Earth's surface. The total amount of carbon in the ocean is about 50 times greater than the amount in the atmosphere, and is exchanged with the atmosphere on a time-scale of several hundred years.



### Anthropogenic Carbon calculations

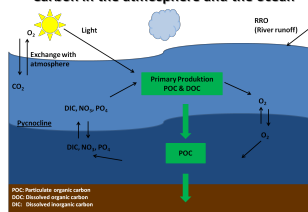
In order to reconstruct the anthropogenic part ( $C_{ant}$ ) of the observed Dissolved Inorganic Carbon ( $DIC_{obs}$ ) in the water samples, it is necessary to derive how much of the carbon/CO<sub>2</sub> has been removed by primary production ( $DIC_{bio}$ ), and how much of the DIC has a natural/preindustrial source ( $DIC_{pi}$ ).

$$C_{ant} = DIC_{obs} - \Delta DIC_{bio} - DIC_{pi}$$

$DIC_{obs}$  is calculated from the chemical parameters Alkalinity, pH and different physical factors and with an atmospheric CO<sub>2</sub>-concentration of 380 ppm. In order to calculate the  $DIC_{pi}$  we used the atmospheric CO<sub>2</sub>-concentration of 280 ppm, which corresponds to preindustrial values, assuming that the physical characteristics (T, S, nutrients) were the same.  $DIC_{bio}$  is an estimation of how much CO<sub>2</sub> has been removed by primary production. This is calculated indirectly by estimating how much oxygen has been used by organisms since the last time the water mass was at the surface, under the assumption that the oxygen saturation of surface water is 100%.

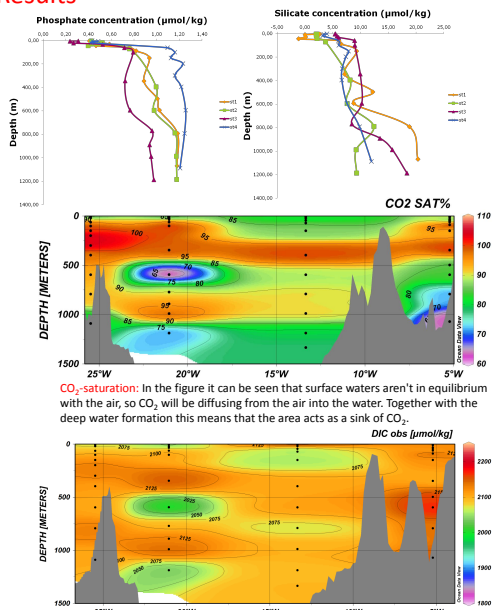
In small timescales  $DIC_{bio}$  is insignificant, because the amount of CO<sub>2</sub> that is removed by water is much larger. In a larger timescale however the  $DIC_{bio}$  has an important role, because the water bound CO<sub>2</sub> returns to the surface again at some time, but some of the  $DIC_{bio}$  will sediment, bury and thereby be removed from the system. [Sarmiento & Gruber, 2006]

### Carbon in the atmosphere and the ocean



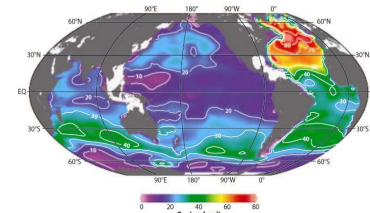
The different Carbon fluxes in the ocean

## Results

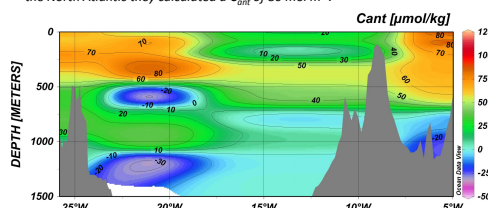


**CO<sub>2</sub>-saturation:** In the figure it can be seen that surface waters aren't in equilibrium with the air, so CO<sub>2</sub> will be diffusing from the air into the water. Together with the deep water formation this means that the area acts as a sink of CO<sub>2</sub>.

**DIC<sub>obs</sub>:** The figure shows  $DIC_{obs}$  (DIC observed), which is higher than the  $DIC_{pi}$ . The difference between the two is around 55-65  $\mu\text{mol kg}^{-1}$ .



Sabine and Tanhua (2010) estimated the oceans anthropogenic carbon. In the North Atlantic they calculated a  $C_{ant}$  of 80  $\text{mol m}^{-2}$ .



$C_{ant}$ : The figure shows  $C_{ant}$  (the anthropogenic carbon).

> The anthropogenic CO<sub>2</sub> values (in Table) that we measured are close to the values that Sabine & Tanhua (2010) presented for the N. Atlantic Ocean, although our values may be underestimated due to the assumptions that we made during our calculations.  
> The negative  $C_{ant}$  values that we observe are due to some extreme values in the physical and chemical measurements (e.g. temperature below 0°C at the first station)

Station	CO <sub>2</sub> anthropogenic [mol/m <sup>2</sup> ]
St1	31,16
St2	24,28
St3	25,11
St4	40,89

## Conclusion

- ✓ The silicate and phosphate vertical profiles show a depletion of nutrients at the surface waters, as expected, due to biological activity.
- ✓ From our CO<sub>2</sub> %-saturation plot we see that the eastern part of the study area is not saturated yet, which means the ocean acts as a sink. On the western part (station 4), however, we observed highly saturated water masses.
- ✓ We also estimated higher DIC concentration today ( $DIC_{obs}$ ) compared to the preindustrial period ( $DIC_{pi}$ ; plot not shown here). We can hence see that anthropogenic CO<sub>2</sub> emission have increased oceanic carbon levels.
- ✓ By comparing our results with results in the literature [Sabine & Tanhua 2010, Gruber 1996] we can see that our values are close to already published data.

## References

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- Sabine, C. L. & Tanhua T. (2010), *Estimation of Anthropogenic CO<sub>2</sub> Inventories in the Ocean*, Annual Reviews.
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<http://www.whrc.org/global/carbon/index.html> (last viewed 14<sup>th</sup> September 2012)

# Oceanography at Sea: Optics

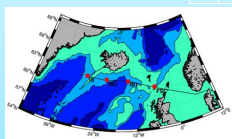
Ciarán Murray<sup>1</sup>, Jakob Thyrring<sup>1</sup>, Mette Vodder<sup>1</sup>, Ida Ringgaard<sup>2</sup>

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## Introduction

The aim of our work was to investigate the optical properties of the North Atlantic Ocean. During a cruise in August and September 2012, measurements were made and water samples were taken from 4 stations: a station in the Faroe Scotland Channel (FSC), 2 stations in the Iceland Basin (IB1, IB2) and a station in the Irminger Sea (IS).

Stn.	Date/Time (2012)	Position
FSC	29 Aug 12:23	60° 21.59' N 5° 22.39' W
IB1	30 Aug 09:01	61° 43.99' N 13° 34.15' W
IB2	31 Aug 08:38	61° 50.66' N 21° 09.10' W
IS	01 Sep 07:41	63° 01.52' N 25° 46.76' W



## Materials and Methods

**I:** During CTD casts, light intensity was recorded every 0.20 meters to give profiles of PAR (photosynthetically available radiation) through the water column. The diffuse light attenuation coefficient,  $K_d$ , was estimated for the surface waters (0-30m) by linear regression on the log-transformed light intensity.

**II:** To estimate chlorophyll specific absorption, water samples from 2 depths (surface and DCM) were filtered onto a Whatman GF/F filter. The optical density of the filters was measured with a Shimadzu UV-2500 spectrophotometer using an integrating sphere attachment. The absorption coefficients were calculated by the following equation:  $a(\lambda) = 2.3 \cdot OD(\lambda) \cdot s / (V \cdot \beta(\lambda))$  where  $\beta$  is a correction factor  $\beta(\lambda) = 1.63 \cdot OD(\lambda)^{-0.22}$  (Bricaud & Stramski 1990).

To calculate the chlorophyll specific absorption, the absorption coefficients were divided by the measured chlorophyll-a concentrations at the respective depths.

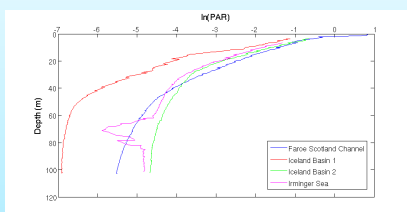
**III:** To investigate the origin of CDOM, water samples were collected from each station. Water was filtered through a 0.2  $\mu\text{m}$  filter, and by using 10 cm quartz cuvettes the samples were analysed in a spectrophotometer. An exponential function was fitted to the observed absorption spectra to give estimates of spectral slope coefficient (S) and absorption coefficient at 375 nm ( $a_{375}$ ):

$$a(\lambda) = a(\lambda_0) \cdot \exp(S(\lambda - \lambda_0)) \quad (\text{Stedmon \& Markager, 2001})$$

**IV:** A photon budget was made for station FSC. Spectral attenuation was modelled by the sum of calculated absorption components and backscattering  $K_d(\lambda) = (a_{\text{water}}(\lambda) + a_{\text{CDOM}}(\lambda) + a_{\text{partic}}(\lambda) + b_b) / \cos(\theta)$ . The backscatter  $b_b$  and zenith angle  $\theta$  were estimated by fitting the model to the observed PAR profile.

## Results and Discussion

### Light attenuation through the water column

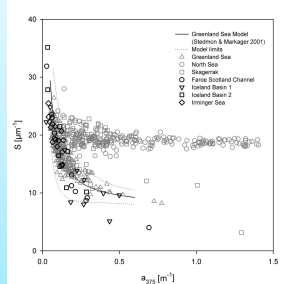


Light attenuation coefficient  $K_d$  (0-30m)

Stn	$K_d$ [ $\text{m}^{-1}$ ]
FSC	0.111
IB1	0.166
IB2	0.121
IS	0.126

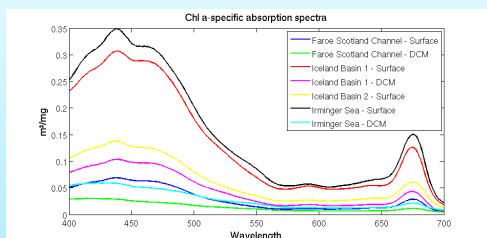
The data from the stations shows decrease in light intensity with depth is approximately exponential over the uppermost 30-40m of the water column. There is a clear reduction in attenuation below the pycnocline. The light attenuation is greatest at IB1.

### CDOM pool comparison



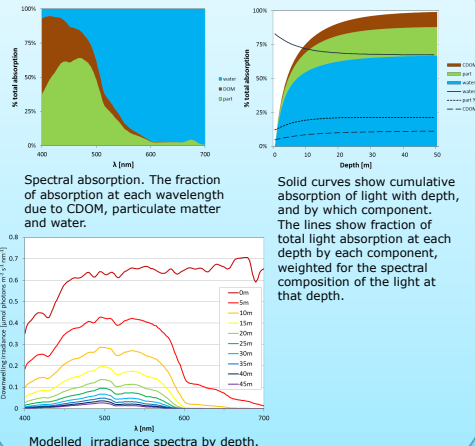
CDOM data from the four stations plotted with data from the Greenland Sea, the North Sea and Skagerrak (Stedmon & Markager, 2001)

### The chlorophyll-a specific absorption spectra



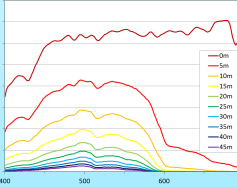
Chlorophyll a specific absorption at all stations. Peaks between 650-700 show absorption of visible red light and peaks around 450 absorption of visible blue light. Surface waters from Irminger Sea and Iceland Basin 1 were found to have the strongest chlorophyll specific absorption: the values are very high in comparison with oceanic (Greenland Sea) values observed by Stæhr et al (2004)

### Where do the photons go?



Spectral absorption. The fraction of absorption at each wavelength due to CDOM, particulate matter and water.

Solid curves show cumulative absorption of light with depth, and by which component. The lines show fraction of total light absorption at each depth by each component, weighted for the spectral composition of the light at that depth.



Modelled irradiance spectra by depth.

## Conclusions

PAR attenuated through the water column and the highest attenuation was seen at Iceland Basin 1. There was no clear relation between chlorophyll-a specific absorption and depth or station, or to overall attenuation. The CDOM seen in the four stations had similar optical characteristics to CDOM from the Greenland sea and most likely derives from in situ production in the ocean.

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# Oceanography at Sea - Physics

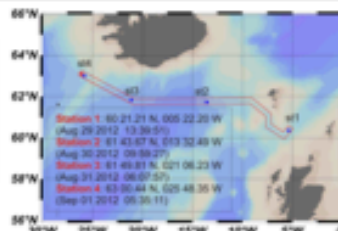
Anette Maria Christensen; Arief Rullyanto; Karl-Søren Geertsen; Søren Enghoff-Poulsen



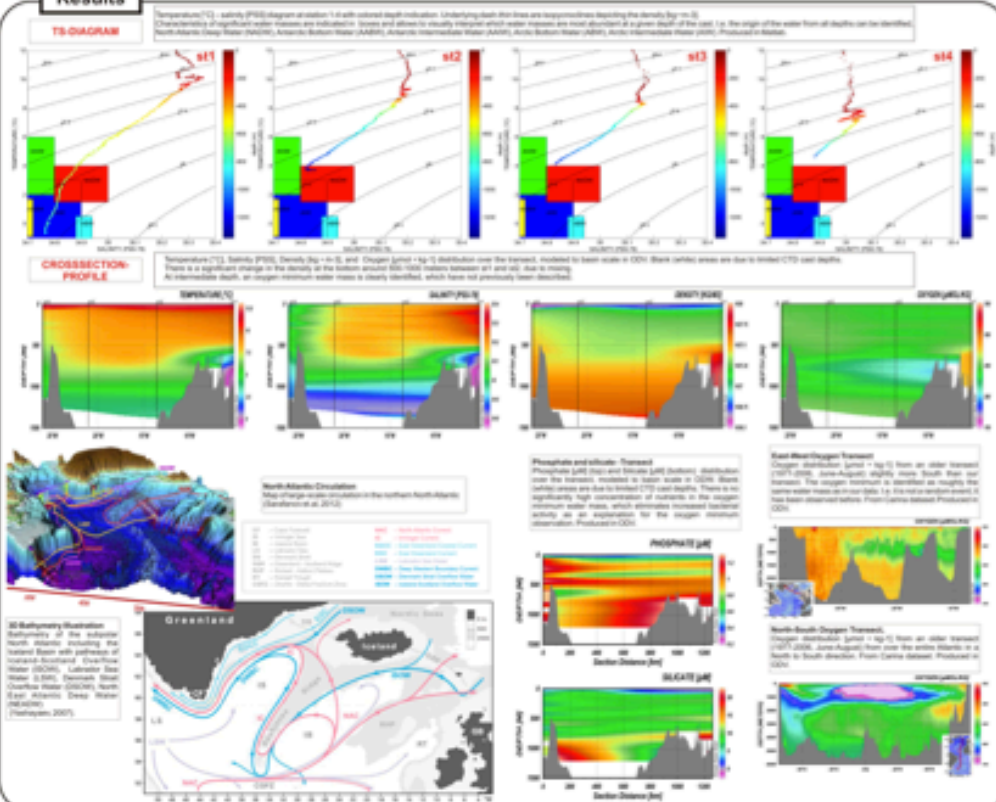
## Abstract

The course Oceanography at Sea was an introductory course to practical aspects of oceanographic research at sea, providing the participating students experience with methods in oceanographic sampling and data analysis in biology, chemistry, optics and physics. The course targeted master- and PhD- students and was carried out on board R/V Dana during a cruise from Hirtshals, Denmark to Reykjavik, Iceland from August 27th to September 2nd 2012.

The focus of this poster is on the physical aspects of the water masses in the North Atlantic just south of the Greenland-Scotland Ridge in the most northern part of the Iceland Basin and the Scotland-Shetland Channel. An area which dynamics have fundamental influence on the world ocean circulation, ecosystems and the climate on Earth. The areas distinct characterization is the cool, saline overflow water from the Danish Strait and the Iceland-Scotland Ridge. Originating from the Nordic Seas (and the Labrador Sea) the various water masses have been identified in below TS-diagrams. A peculiar water mass with distinct low oxygen content was observed at the eastern part of the transect at depths from 600-800 m. Older datasets for this area observes the same feature and with no mixing at these depths, which suggest the oxygen minimum feature to be permanent. We conclude the water to be relatively low in oxygen due to water mass aging and hypothesize the water mass to originate from Antarctic Intermediate Water and/or North-Atlantic Deep Water.



## Results



## Summary and Conclusion

The results show a water mass with distinct low oxygen content in the eastern part of the transect at depths from 600-800 m. Online available datasets (Glodap 1972-1999; WOCE 1990-2009; Carina 1977-2006) and models (Monteyri and Levitus 1997) for this area shows the same pattern in vertical oxygen distribution even at winter (Glodap 1972-1999) and the water does not seem to be mixed at these depths, which suggest the oxygen minimum feature to be permanent. At the same depths the nutrient content derived from the sampling at this cruise was not significantly higher than the surrounding water masses which eliminates increased, local bacterial decomposition to explain the observation. Due to the non-significant relation to nutrient content and the water being relatively low in oxygen, we conclude that the water mass is old water with oxygen consumed through many years of respiration, and hypothesize the water mass to originate from Antarctic Intermediate Water and/or North-Atlantic Deep Water. Older datasets observe the oxygen minimum water mass at deeper depths, which we hypothesize owes to the their more southern transects in the Iceland Basin, where the relative density to the surrounding water is thus higher which situates the water mass at deeper depths. Due to the seemingly shortage of winter datasets for this region (only one available dataset, Glodap, with data going back to 1972-1999) we suggest future oceanographic cruises to supply this information to confer if the oxygen minimum water mass indeed is not mixed with the upper water masses by deep convection. If the mentioned water mass turns out to be absent at winter, the water column can be concluded to be mixed by convection and/or chimney effects. To at least these depths. This allows for a basin scale calculation of the water volume mixed, which provide information on how much CO<sub>2</sub> is getting pumped down in the Iceland Basin by assuming the oxygen being replaced by CO<sub>2</sub> when in contact with the atmosphere. The Iceland Basin scale in mind this potential is quite substantial and potentially has influence on the area as a functioning carbon sink.